

THE STRUCTURE AND ABSOLUTE CONFIGURATION OF PLENOLIN,

A CYTOTOXIC SESQUITERPENE LACTONE

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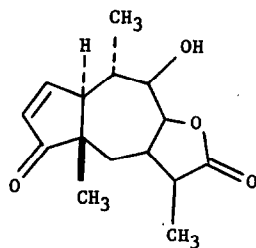
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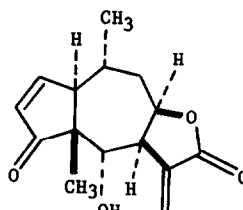
We had previously reported¹ the isolation of a new sesquiterpene lactone, plenolin, from Baileya pleniradiata Harv. and Gray in 0.0003% yield. Too little plenolin was obtained to permit detailed chemical investigation although the structure I was tentatively assigned on the basis of limited spectral data.

While searching for a supply of helenalin (II) for the investigation of the relationship between the sesquiterpene lactone structure and the antitumor/cytotoxic activity,² we isolated plenolin as a cytotoxic principle^{3,4} from Florida Helenium autumnale L.^{5,6} and are reporting our reinvestigation of the structure of plenolin.⁷

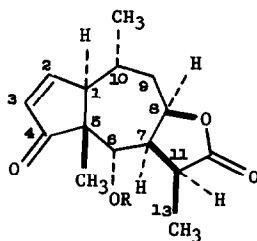
Plenolin was isolated in 0.13% yield from the mother liquor after the removal of helenalin by repeated silica gel column as well as preparative chromatography. Plenolin (III), m.p. 223-226°, ⁸ has the composition C₁₅H₂₀O₄ and shows the presence of a hydroxyl group (3450 cm⁻¹) and a γ -lactone ring (1745 cm⁻¹). The presence of a cyclopentenone ring system in plenolin is indicated by the appearance of an IR band at 1712 cm⁻¹ and is substantiated by the presence in the NMR spectrum (in pyridine-d₅) of the characteristic doublets of doublets at δ 7.60 (H-2, J = 2, 6 Hz)⁹ and 6.08 (H-3, J = 3, 6 Hz). Plenolin lacks the α -methylene group on the lactone ring, and is the corresponding α -methyl- γ -lactone. Three-proton methyl group signals are seen at δ 1.26 (C-11 methyl, d, J = 7 Hz), 1.08 (C-5 methyl, s) and 1.02 (C-10 methyl, d, J = 7 Hz). The lactonic proton at C-8 is seen as a multiplet at δ 4.87. The one proton doublet¹⁰ at δ 4.73 (J = 4 Hz) is assigned to the proton at the hydroxyl group (at C-6) since it is shifted downfield to δ 5.48 (s) upon acetylation with acetic anhydride in pyridine.



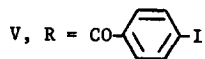
I



II



III, R = H

IV, R = COCH₃V, R = CO-C₆H₄-I

Phenolin acetate (IV), C₁₇H₂₂O₅, has m.p. 174-175° and exhibits NMR (CDCl₃) signals at δ 1.05 (3H, s, C-5 methyl), 1.25 (3H, d, J = 7 Hz, C-10 methyl), 1.51 (3H, d, J = 7 Hz, C-11 methyl), 1.94 (3H, s, acetyl), 4.78 (1H, m, H-8), 6.10 (1H, dd, J = 3, 6 Hz, H-3) and 7.75 (1H, dd, J = 2, 3 Hz, H-2).

Considerations from the biogenetic implications observed in the co-occurrence of plenolin (III) and helenalin (II), coupled with the evidence described above, led to the conclusion that the structure of plenolin must be modified to III, i.e. 11,13-dihydrohelenalin. To confirm further the structure of plenolin, the identities of III and IV were established by direct comparison (mixed m.p.; tlc; superimposable IR and NMR spectra) with synthetic samples of dihydrohelenalin and acetyldihydrohelenalin prepared by catalytic hydrogenation of helenalin and acetylation of dihydrohelenalin.

In order to define the stereochemistry at C-11, a single crystal X-ray analysis of the 6-*p*-iodobenzoate derivative (V) was undertaken. The crystals belong to the orthorhombic system, space group $P2_12_12$, with four molecules in a unit cell of dimensions $a = 9.29$, $b = 26.85$, $c = 8.45$ Å. The structure was solved by the heavy-atom approach and refined by full-matrix least squares calculations incorporating the anomalous scattering corrections for the iodine atom to $R = 0.054$ over 1165 reflexions obtained from diffractometer measurements. The analysis shows that V represents the structure and absolute stereochemistry of the *p*-iodobenzoate derivative from which it follows that III represents dihydrohelenalin, i.e. plenolin.

The potent cytotoxicity exhibited by plenolin is in agreement with our previous hypothesis,^{4,11} i.e. the structural requirement for significant cytotoxicity is an $O=C-C=CH_2$ system as an active center in a lactone, or in a ketone, especially when it is in a cyclopentenone ring system.

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REFERENCES AND FOOTNOTES

1. T. G. Waddell and T. A. Geissman, Phytochemistry, **8**, 2371 (1969).
2. K. H. Lee, J. Pharm. Sci., **62**, 1028 (1973), and references cited therein.
3. Plenolin showed significant inhibitory activity of the in vitro growth of tissue culture cells originating from human epidermoid carcinoma of larynx (H.Ep.-2) at 0.814 mcg/ml.⁴
4. K. H. Lee, H. Furukawa, and E. S. Huang, J. Med. Chem., **15**, 609 (1972).
5. Specimens were gathered in October, 1971 in Jackson County, Florida. The constituents of Florida H. autumnale was previously examined and reported to contain helenalin in good yield.⁶
6. W. Herz, A. Romo de Vivar, J. Romo, and N. Viswanathan, J. Amer. Chem. Soc., **85**, 19 (1963).
7. Plenolin isolated either from H. autumnale or B. pleniradiata was found to be identical on the basis of direct comparison (mixed m.p.; tlc; superimposable IR and NMR spectra).
8. Lit. 1 reported m.p. 216-220°.
9. Partially overlapped with the signal of pyridine.
10. This doublet became a singlet upon addition of D₂O or acetylation indicating the coupling of H-6 with the hydroxyl proton.
11. K. H. Lee, E. S. Huang, C. Piantadosi, J. S. Pagano, and T. A. Geissman, Cancer Res., **31**, 1649 (1971).